# Introduction to Macromolecular Crystallography

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# 2014 - International Year of Crystallography



developing world. Students from hearby countries will trave to attend tutorials, workshops and The hubs will also host conferences and exhibitions. find out more at

Crystal-growing competition for schools Fellowship initiatives in Africa, Latin America and South-East Asia

Exhibitions, public engagement events

www.iycr2014.org

- 1895: Discovery of X-rays by Wilhelm Conrad Roentgen
- 1912: Discovery of diffraction of X-rays by crystals by Max von Laue, Paul Knipping and Walter Friedrich
- 1913: William Lawrence Bragg and William Henry Bragg propose Bragg's law

# **Electromagnetic Radiation**



### **Protein Crystallography - Overview**



### **Diffraction Experiment**



X-rays: high energy electromagnetic radiation (wavelength ~ 1 Å)

Electrons scatter X-rays (amplitude and phase)

Crystals contain atoms in a defined arrangement

Scattered waves interfere (constructive vs. destructive interference)

Interference pattern depends on arrangement of atoms

Resolution limit: up to 1 Å

Virtually no size limit

We cannot measure the phases of the diffracted X-rays!

### **Diffraction Experiment vs. Light Microscope**



# **Crystal Growth**



Salting out with  $(NH_4)_2SO_4$ : Solubility of myoglobin is reduced

Trial and error process



# Crystals



Colored due to heme prosthetic group (usually colorless)

Size: > 0.1 mm in each direction

Fragile

Prone to dehydration: 30%-80% solvent content

Soaking experiments

### **Crystal Mounting**



Capillary-mounted crystal for room temperature data collection



Loop-mounted crystal for cryogenic data collection

### Asymmetric Unit, Unit Cell and Crystal



## X-ray generator



### Rotating anode generator

Micromax 007-HF 70 x 70 μm focal spot 9000 rpm 1.2 kW (40 kV and 30 mA)





**Rotating Anode** 

Filament

### X-ray spectra



Continuous X-ray spectra as a function of accelerating voltage  $\lambda_{min}$  = 12,398 / V<sub>acc</sub> (Å)

X-ray spectra with characteristic peaks (Mo at 50 kV and Cu at 35 kV)

$$\begin{array}{ll}\mathsf{K}_{\alpha 1},\,\mathsf{K}_{\alpha 2} & \mathsf{L} {\rightarrow}\,\mathsf{K}\\ \mathsf{K}_{\beta 1},\,\mathsf{K}_{\beta 2} & \mathsf{M} {\rightarrow}\,\mathsf{K}\end{array}$$

### Synchrotrons I

#### French Alps



#### Main Advantages

Higher intensity Microfocus Time-resolved Laue crystallography

Tunability MAD/SAD

Ľlsère

How does it work?

Storage Ring

# Synchrotrons II



#### **Focussing Magnet:**

Placed in the straight sections of the storage ring

Focus electron beam to keep it small and well-defined

A small and well-defined electron beam will produce the very bright X-ray beam needed

#### **Bending Magnet:**

Electrons are deflected in BM from their straight path by several degrees

Change in direction causes electrons to emit synchrotron radiation

#### **Undulators:**

Complex array of small magnets

Force electrons to follow an undulating, or wavy, trajectory

Beams of radiation emitted from different bends overlap and interfere with each other to generate a much more intense beam

### **Diffraction Pattern**



Recorded on hybrid pixel array detectors (old: Xray sensitive film, imaging plates, CCD detectors)

Multiple images give a complete dataset

Crystal is exposed in different orientations

Consists of 10<sup>4</sup>-10<sup>6</sup> spots, which differ in their intensity.



Constructive interference if extra distance is an integral (n) multiple of the wavelength ( $\lambda$ ) :

 $\label{eq:linear} \begin{array}{l} n\ \lambda = AB + BC = 2\ AB \\ \mbox{Remembering the definition of the sine function} \\ AB = d\ sin\theta \\ \mbox{Combining the two equations yields Bragg's law:} \\ n\ \lambda = 2\ d\ sin\theta \end{array}$ 

## A Lattice and its Diffraction Image



### **Representation of Phases and Amplitudes**



Amplitude is represented by color saturation and brightness (small amplitudes: faded colors, large amplitudes: saturated colors)
Phase is given by hue.
The positive real (horizontal) axis to the right represents a phase of 0.
Phase increases anti-clockwise

0 degrees is red
120 degrees is green
240 degrees is blue.

### **From Molecules to Crystals**



### **Duck Tales**



The diffraction pattern of any object can be computed.

## **Animal Magic I**



### A duck and its Fourier Transform

A cat and its Fourier Transform

# **Animal Magic II**



Duck magnitudes and cat phases reveal a: Cat

Cat magnitudes and duck phases reveal a: Duck

### **Structure Factor and Electron Density Equations**

$$\mathbf{F}(\mathbf{h}) = \int_{cell} \rho(\mathbf{x}) \exp(2\pi i \mathbf{h} \cdot \mathbf{x}) d\mathbf{v} \qquad \rho(\mathbf{x}) = \frac{1}{V} \sum_{\mathbf{h}} \mathbf{F}(\mathbf{h}) \exp(-2\pi i \mathbf{h} \cdot \mathbf{x})$$

**h** is the vector (h,k,l) and **x** is the vector (x,y,z)  $\mathbf{h} \cdot \mathbf{x} = hx + ky + lz$ 

Both operations are Fourier transformations, which are mathematical vehicles between real space ( $\rho(\mathbf{x})$ ) and reciprocal space ( $F(\mathbf{h})$ )

 $\rho(\mathbf{x})$  has units of electrons per Å<sup>3</sup> and  $\mathbf{F}(\mathbf{h})$  of electrons. Specifically, F(000) is equal to the number of electrons in the unit cell. F(000) cannot be measured since it coincides with the primary beam

Fourier transformation of the electron density yields the structure factor and Fourier transformation of the structure factor yields the electron density

These operations are sometimes also referred to as Fourier synthesis and Fourier analysis

### **Resolution I**







### **Electron Density Map**



X-rays are diffracted by electrons

3D Map in general, here shown as a 2D slice

Higher contour levels for more electron dense atoms (Fe)

Atomic model has to be fitted to electron density map

### **Resolution II**



Electron density maps at (A) 4 Å, (B) 2.5 Å and (C) 1 Å resolution. In the 1 Å map there is a low contour level in blue and a high contour level in red.

# **Quality of Refinement**

Refinement progress has been traditionally monitored by the R-factor:

$$R = \sum_{hkl} ||F_o| - |F_c|| / \sum_{hkl} |F_o|$$

Typical R-values for macromolecular structures are ~0.2 (20%).

For a set of randomly distributed atoms in an acentric crystal: R = 0.586

R can be lowered artificially by refining additional parameters (B-factors when not appropriate, addition of spurious solvent molecules).

More recently the free R-factor (R<sub>free</sub>) has been used, in which the R-factor is calculated for a subset of reflections which are never included in the refinement.

R<sub>free</sub> is sensitive towards overfitting, i.e. the refinement of parameters not warranted at a given resolution.

Typical R<sub>free</sub>-values for macromolecular structures are ~0.25-0.3 (25-30%).

## The Final Model

Each crystal structure will result in a file containing the coordinates of all atoms together with their B-factor and occupancy.

Various graphic programs are available to display these models.

Two particularly useful programs are: Swiss-PDBViewer: <u>http://ca.expasy.org/spdbv/</u> Pymol: <u>http://www.pymol.org/</u>

The model will be analyzed to understand the biological function of the macromolecule.

Macromolecular structures are deposited in the Protein Data Bank: <u>http://www.rcsb.org</u> or <u>http://www.ebi.ac.uk/pdbe/</u>

The Protein Data Bank can be searched to retrieve entries of interest.

### **Image Sources**

J.A. Gatehouse

Jeremy M. Berg, John L. Tymoczko and Lubert Stryer "Biochemistry", W.H. Freeman

Rigaku - X-ray Analytical Instrumentation <u>www.rigaku.com</u>

European Synchrotron Radiation Facility (ESRF) <u>www.esrf.eu</u>

Kevin Cowtan's home page: <u>http://www.ysbl.york.ac.uk/~cowtan/fourier/fourier.html</u>